

# Laboratory Spectral Simulations of Exoplanet Atmospheres

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Program: Strategic Initiatives

## Project Objective:

The objectives of our laboratory investigations are to:

1. Produce high resolution infrared spectroscopy data for equilibrium atmospheres
2. Follow photochemistry and aerosol formation in hot-Jupiter-type exoplanet atmospheres as they evolve under high temperatures and UV radiation

Our work will help understanding the role of UV-photochemistry in exoplanet atmospheres in addition to thermochemistry.

## FY18/19 Results:

- Studied thermochemistry in  $\text{H}_2:\text{CO}$  (300:1) and  $\text{H}_2:\text{CO}:\text{H}_2\text{O}$  (400:1:1) atmospheres using the **CAAPSE (Cell for Atmospheric and Aerosol Photochemistry Simulations of Exoplanets)** experiment from 1173 K to 1473 K and the photochemistry using UV-radiation at Ly- $\alpha$  (121.6nm).
- Formation of  $\text{CO}_2$  in the gas-phase with thermochemistry
- Formation of  $\text{CO}_2$  and  $\text{H}_2\text{O}$  in the gas-phase with photochemistry with  $\text{H}_2\text{O}$  being more important at  $T > 1200$  K
- Determine the source of oxygen to be CO for the formation of  $\text{CO}_2$  under UV irradiation

## 3. Approach:

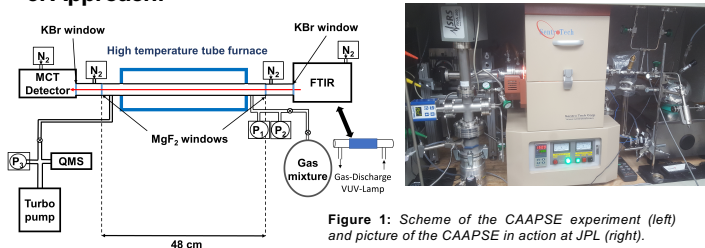


Figure 1: Scheme of the CAAPSE experiment (left) and picture of the CAAPSE in action at JPL (right).

The CAAPSE setup is presented in Figure 1. The cell is filled with 15 mbar of  $\text{D}_2:^{13}\text{CO}$  (300:1) or  $\text{D}_2:^{13}\text{CO}:\text{D}_2\text{O}$  (400:1:1) gas mixtures to simulate hydrogen-dominated atmosphere with C/O ratios of 1 and 0.5. The gases are heated at temperatures up to 1473 K. When the gaseous mixture has reached the thermal equilibrium composition, it is irradiated using a microwave discharge hydrogen flow lamp (Ly- $\alpha$  (121.6 nm) emission band plus a broad emission band in the 140–170 nm range, mimicking the VUV solar flux received in simulated exoplanet atmospheres).

## 4. Source of Oxygen for the formation of $\text{CO}_2$ in $\text{H}_2/\text{CO}$ atmospheres:

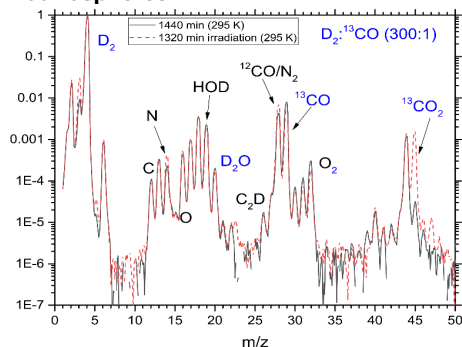


Figure 2: Evolution of the gas phase composition after 1320 min of UV irradiation of  $\text{D}_2:^{13}\text{CO}$  (300:1) gas mixture at room temperature, monitored with in situ mass spectrometry. These results showed that for two  $^{13}\text{CO}$  molecules consumed, one  $^{13}\text{CO}_2$  molecule is formed, while  $\text{H}_2\text{O}$  or  $\text{O}_2$  depletion observed cannot account for  $^{13}\text{CO}_2$  formed. These results highlight that CO is the source of oxygen in the photochemical formation of  $\text{CO}_2$ .

- We shown that  $\text{CO}_2$  could be formed photochemically in  $\text{H}_2:\text{CO}$  atmospheres (Fleury et al. 2019)
- Figure 2 shows the mass spectra of the  $\text{D}_2:^{13}\text{CO}$  gas mixture at 295 K before and after irradiation. We observed the consumption of  $^{13}\text{CO}$  ( $m/z$  29) and the formation of  $^{13}\text{CO}_2$  ( $m/z$  45).
- We found that CO is the source of oxygen for the formation of  $\text{CO}_2$ .

## 5. Thermochemistry and Photochemistry in $\text{H}_2/\text{CO}/\text{H}_2\text{O}$ atmospheres:

Figure 3 shows the IR-Absorption spectra (irradiated and unirradiated) in the 1800-3000  $\text{cm}^{-1}$  (3.3-5.5  $\mu\text{m}$ ) range with a resolution of 0.25  $\text{cm}^{-1}$  at different temperatures.

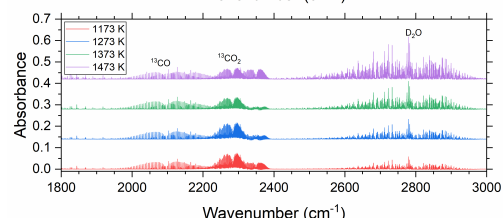
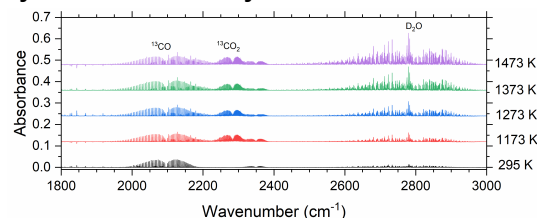


Figure 2: IR spectra of the initial gaseous mixture of  $\text{H}_2:^{13}\text{CO}$  (99.7%:0.3%) at 295 K and at the equilibrium for different set oven temperatures: 573 K, 873 K, 1173 K, 1273 K, 1373 K, and 1473 K (top) and after 18 hours of irradiations at 121.6 nm (Ly- $\alpha$ ) (bottom); spectra are offset for clarity.

The room-temperature spectrum contains the absorption band of the initial  $^{13}\text{CO}$  (2095  $\text{cm}^{-1}$ ) and  $\text{D}_2\text{O}$  (2780  $\text{cm}^{-1}$ ).

The spectra of the heated gases reveal the formation of bands due to two new (thermally-generated) molecular species:  $^{13}\text{CO}_2$  (2284  $\text{cm}^{-1}$ ) and  $\text{D}_2\text{O}$  (2780  $\text{cm}^{-1}$ ).  $\text{CO}_2$  is isotopically labelled with  $^{13}\text{C}$ , highlighting that it is formed during thermal chemistry from the initial  $^{13}\text{CO}$ .

UV (121.6 nm, Ly- $\alpha$ ) irradiation of the heated gases leads to a decrease of  $[\text{CO}]$ , an increase of  $[\text{CO}_2]$ , and the formation of water. Photochemistry can strongly influence the composition of  $\text{H}_2:\text{CO}:\text{H}_2\text{O}$  warm exoplanet atmospheres.  $\text{CO}_2$  and  $\text{H}_2\text{O}$  can be efficiently produced in these atmospheres despite competitive loss processes. The gas temperature strongly influences the efficiency of the different chemical pathways, notably due to the increase of the absorption cross-sections of  $\text{CO}_2$  when the temperature increases.

## Benefits to NASA and JPL (or significance of results):

These results highlight the importance of including UV photochemistry in models used to simulate exoplanet atmospheres. Photochemistry and thermochemistry are both heavily involved in the composition of hot-Jupiter-like exoplanet atmospheres, and these results can be used to better interpret ground- and space-based spectroscopic observations of these exoplanet atmospheres. For missions like JWST, our studies will provide much-needed laboratory reference data for atmospheric molecules and aerosols in the infrared and ultraviolet spectral regions.

## Publications:

B. Fleury, M. S. Gudipati, B. L. Henderson, and M. Swain, "Photochemistry in hot  $\text{H}_2$ -dominated exoplanet atmospheres", (2019), ApJ 871:158

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